# Real-Time Video Transmission for Underwater Wireless Optical Communications System

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### Abstract

In this paper, it designed and experimentally demonstrate an underwater wireless optical communication system by using a multi-beam green laser diode. The system was capable of transmitting a data video with pulse width modulation (PWM). Two laser diode (532 nm) used to carry the data (video) and two photodiode utilities as a receiver. It used MgSO<sub>4</sub> and Mg(OH)<sub>2</sub> as a suspended particles. The main purpose of adding these compounds is to a simulation of the water in nature. Also to find the difference between the affect of adding each chemical compounds to clean water on the transmitting of data (video). The gain, loss, SNR and BER have calculated for each one. It is found that MgSO<sub>4</sub> affect was greater than Mg(OH)<sub>2</sub> on the received video signal.

### I. Introduction

The importance of underwater wireless optical communications (UWOC) is becoming increasingly because it has high transmission rates. For short- and medium-range applications, wireless optical technology is usually trusted solution under water [1]. Undiscovered underwater radio channels are still more complex than free satellite channels and scientific problem of underwater communications across a range of physical processes in different types of underwater environments ranging from coastal waters to deep sea or oceans [2].

The most important characteristic of laser in underwater optical communications is the possibility of information transmitting through it at a higher speed and better accuracy than sound and radio waves [3]. The delay in the propagation of sound waves in water is more than five times of radio frequencies. In addition to that, sound waves propagate much slower under water (1500 m / s) than electromagnetic waves [3].

### **II. Theoretical Background**

The attenuation mainly represented by absorption and scattering, which are the main cause of light loss in water [4]. It is obviously that increasing the turbidity of clean water decreases reception power [5,6]. Due to that attenuation which emerges from absorption and scattering in the water. Best transmissions are achieved in clean water because less attenuation occurred due to less absorption and scattering. Larger attenuation is occurred for high concentrations of salts [7].

The absorption by water itself is known, but the absorption by different dissolved particles need to study. The concentrations of these dissolved particles is variable. The spectral absorption of water is a combination of the absorption by pure sea water, phytoplankton, detritus and Colored Dissolved Organic Matter (CDOM).

#### **III. Experimental Setup**

The wireless optical communication system using multi-beams as shown in figure (1). The data (video) transmitted from the computer (CP) to the Transistor-Transistor-Logic (TTL) (Model: CP2102) circuit which is the process of modulation and then transferred to the laser sources (two green lasers with wavelength 532nm). the laser signal travelling through a glass

water tank with dimensions  $(100 \times 40 \times 30)$  cm<sup>3</sup>, and filled with (80 liters) of clean water. It has been added Mg(OH)<sub>2</sub> and MgSO<sub>4</sub>, to the clean water for increasing the turbidity.

The added concentrations of compounds to the water shown in table in (I).

compounds	Concentrations (mg)								
MgSO <sub>4</sub>	25	50	75	100					
Mg(OH) <sub>2</sub>	25	50	75	100	125	150	175	200	225

Table (I) the added concentrations of compounds to the water

The transmitter signal received by using pair of photodiodes (BPX65), which work to convert the signal into the TTL circuit. Where the TTL circuit included the process of demodulation the signal and the reception video transferred to the computer (CP) to observe by the screen.



Fig (1) Shows Experimental setup for underwater communication system.

A digital storage oscilloscope (DSO) used to display the received signal waveforms. The power received for each source was measured using the Power Meter.

A 25 mg of  $Mg(OH)_2$  and  $MgSO_4$  added to water in the tank. The mixture was mixed to get the homogeneity of chemical compounds with the water. It have

increasing the concentration of the compounds by adding the compounds step by step.

Loss (L) in dB calculated by using eq (1) [8].

$$L_{dB} = 10 \log_{10} \frac{P_{in}}{P_{out}}$$
(1)

Signal to noise ratio (SNR) has calculated based on DSO measurement. Equation (2) employed to calculate SNR [9] depending on the maximum and minimum amplitude of voltage in the receiver part.

$$SNR(dB) = 20 \log_{10} \frac{High - Low}{Low}$$
 (2)

Where High is the measured amplitude value at the photodiode when the  $LD_s$  are on (signal with noise), Low is the condition when the  $LD_s$  are off (noise only).

Taking advantage of the SNR values under study that were calculated from an equation (2), the Bit Error Rate (BER) values were calculated based on an equation (3) [10]:

$$BER \approx \left(\frac{2}{\pi SNR}\right) e^{-\frac{SNR}{8}}$$
 (3)

#### **IV. Result and Discussion**

The loss is calculated for clean water which has values (1.747, 1.797)dB for photodiode1 (Rx<sub>1</sub>) and photodiode2 (Rx<sub>2</sub>) respectively. For Gain value of clean water was (0.5724, 0.4923)dB for Rx<sub>1</sub> and Rx<sub>2</sub> respectively.

Fig (2) shows the Gain and Loss of clean water with adding of MgSO<sub>4</sub>. When MgSO<sub>4</sub> added to the water the turbidity begins to increase. This added leads

to increase in loss, with increasing concentrations of MgSO<sub>4</sub>.On the other hand side, Gain decrease when the concentrations of MgSO<sub>4</sub> is increasing.



Fig (2) Gain & Loss for MgSO<sub>4</sub>: (a) 1<sup>st</sup> beam. (b) 2<sup>nd</sup> beam.

Fig (3) shows the Gain and Loss of clean water with adding of  $Mg(OH)_2$ . When  $Mg(OH)_2$  added to the water the turbidity begins to increase. This added leads to increase in loss, with increasing concentrations of  $Mg(OH)_2$ .On the other hand side, Gain decrease when the concentrations of  $Mg(OH)_2$  is increasing.



Fig (3) Gain & Loss for  $Mg(OH)_2$ : (a)  $1^{st}$  beam. (b)  $2^{nd}$  beam.

From Figure (2) it is note that the maximum loss were (2.690, 2.652)dB for Rx<sub>1</sub> and Rx<sub>2</sub> respectively to MgSO<sub>4</sub> at concentration 100mg. While the maximum loss were (4.274, 4.152)dB for Rx<sub>1</sub> and Rx<sub>2</sub> respectively for Mg(OH)<sub>2</sub> at concentration 225mg.

As for Gain , it is note that the minimum Gain were (0.3717, 0.3770)dB for Rx<sub>1</sub> and Rx<sub>2</sub> respectively to MgSO<sub>4</sub> at concentration 100mg. While the minimum gain were (0.2339, 0.2408)dB for Rx<sub>1</sub> and Rx<sub>2</sub> respectively for Mg(OH)<sub>2</sub> at concentration 225mg.

This means that less gain was obtained for  $Mg(OH)_2$  compared to  $MgSO_4$ . Therefore  $MgSO_4$  is more gain than  $Mg(OH)_2$ .

Figure (4) shows the SNR for MgSO<sub>4</sub>. It is observed SNR was (22.278, 21.938 , 22.110)dB for both photodiods ( $Rx_s$ ) , first ( $Rx_1$ ) ,and second photodiods ( $Rx_2$ ) respectively for clean water. These values have declined to (20.319,20,20.214)dB for  $Rx_s$ , $Rx_1$ ,and  $Rx_2$  respectively at the adding of concentration (100mg).



Figure (4) Signal to noise ratio for MgSO<sub>4</sub>

Also SNR was calculated for  $Mg(OH)_2$  as shown in fig (5). It is obvioused that SNR was (22.278, 21.938, 22.110)dB for  $Rx_s$ ,  $Rx_1$ , and  $Rx_2$  respectively for clean water. The SNR for  $Mg(OH)_2$  decreased to (19.204,19.204,19.204)dB for  $Rx_s$ ,  $Rx_1$ , and  $Rx_2$  respectively for the adding concentration at (225mg).



Figure (5) Signal to noise ratio for Mg(OH)<sub>2</sub>

The effect of the increase turbidity on the quality of received video were studied as shown figures (6-7) where some of video images received for pure water and in the case of some additions for  $MgSO_4$  and  $Mg(OH)_2$  under study. It is noticed a clear deformation when the turbidity increases by increasing the concentration of added chemical compounds.



(a) (b) (c)Figure (6) shows received videos :(a) for clear water. (b) For 25mg of MgSO<sub>4</sub>. (c) For 75 mg of MgSO<sub>4</sub>.



Figure (7) shows received videos :(a) for clear water. (b) For 25mg of Mg(OH)<sub>2</sub>. (c) For 75 mg of Mg(OH)<sub>2</sub>.

Also it is observed that  $MgSO_4$  has more effect on the laser beams compare with  $Mg(OH)_2$ .

It is noted that the concentration of MgSO<sub>4</sub> effect more than Mg(OH)<sub>2</sub> and required to reach the concentration of 225 for Mg(OH)<sub>2</sub> to obtain the deformation as shown in the figure (8), where figure (8-a) shows received video for 100 mg of MgSO<sub>4</sub>. As for the figure (8-b) shows received video For 225 mg of Mg(OH)<sub>2</sub>.



Figure (8) show received video for: a. 100 mg of  $MgSO_{4.}$  b. 225 mg of  $Mg(OH)_2$ 

It is observed that BER was (0.00176, 0.00186, 0.00181) for  $(Rx_s)$ ,  $(Rx_1)$ , and  $(Rx_2)$  respectively for clean water. These values have increased to

(0.002471, 0.002612, 0.002516) for  $(Rx_s)$ ,  $(Rx_1)$ , and  $(Rx_2)$  respectively at the adding of concentration (100mg) as shown in figure (9).



Figure (9) shows BER for MgSO<sub>4</sub>.





Figure (11) shows BER for Mg(OH)<sub>2</sub>

### **V.** Conclusion

It is shown that the effect of  $MgSO_4$  was greater than the effect of  $Mg(OH)_2$ . This means that the effect of suspended substances represented by  $MgSO_4$ , is greater than the effect of suspended substances by  $Mg(OH)_2$ . Therefore,  $Mg(OH)_2$  has an advantage over  $MgSO_4$  for the transfer of video where the turbidity for  $Mg(OH)_2$  is less compared to  $MgSO_4$ .

It is obvious that the received signal suffers from the attenuation by  $MgSO_4$  compared with  $Mg(OH)_2$ . Also, noted that, the received image of video has more high resolution for water turbidity compound  $Mg(OH)_2$  in water. Beside, low error in the received data (video) reception when the concentration of suspended particles is decreasing.

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